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Refrigeration And Air Conditioning Stoecker Solution Manual

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3-1 Calculate the specific volume of an air-vapor mixture in cubic meters per kilogram of dry air when the following conditions prevail: $t = 30$ C, $W = 0.015$ kg/kg, and $p = 90$ kPa. Solution: Equation 3-4. $t_s = t + 273 = 30 + 273 = 303$ K $R_a = 287$ J/kg.K $P = 90$ kPa = 90,000 Pa Equation 3-2 $t_s = 30 + 273 = 303$ K $W = 0.015$ kg/kg $\rho = 1.35 - 0.15p$ s

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Solution: $q = mc_p(t_2 - t_1)$ $m = 2.5$ kg/s $c_p = 1.0$ kJ/kg.K $t_2 = 30$ C $t_1 = -10$ C CHAPTER 2 - THERMAL PRINCIPLES Page 2 of 5 Then, $q = (2.5)(1.0)(30 + 10)$ $q = 100$ kw - - - Ans. 2-3.

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